CH 111: Introduction to Chemical Principles (4 credits)
University of Oregon, Department of Chemistry and Biochemistry
Fall 2016

Instructor: Professor Brandi Baldock, Ph.D.
Office: 160A KLA
Office phone: (541) 346-2592
Email: bdiickins@uoregon.edu

Lectures: Mondays & Wednesdays 12-1:20pm and Fridays 12-12:50pm in 123 PAC

Office Hours: Students are strongly encouraged to ask questions and seek help early. Chemistry is challenging. Office hours provide a great venue for working on homework problems in smaller groups or to get help on specific topics.

Office Hours: Mon. 4:00 – 4:50 pm, 107 KLA
Wed. 9:00 – 9:50 am, 107 KLA
Fri. 1:00 – 1:50 pm, 107 KLA

Teaching Assistants: Donald Clayton Email: drc@uoregon.edu
Alan Moghaddam Email: alanm@uoregon.edu

GTF Office Hours:
Mon. 2:00-2:50pm 107 KLA Alan Moghaddam
Tues. 5:00-5:50pm 107 KLA Donald Clayton
Wed. 10:00-10:50am 107 KLA Alan Moghaddam
Thurs. 5:00-5:50pm 107 KLA Donald Clayton

Required Materials:
Simple scientific calculator with no data storage
i>clicker 2 (register online)
Online homework program: Mastering Chemistry
Text: Introductory Chemistry Essentials by Nivaldo J. Tro. This text is accessible online within the Mastering Chemistry homework program. Purchase of a looseleaf printed version of the text is optional.

Online Information: Important course information will be posted on Canvas and sent via email. Students are responsible for checking the course Canvas homepage and their UO email daily to avoid missing important course information. The course Canvas website will have a gradebook containing student scores on individual assignments, but the percentage listed will not accurately reflect students’ overall grade in the course.

Pre-requisites: Math 95. Students are expected that to be proficient with exponents, scientific notation, fractions, and solving simple algebraic equations.

Course Content: CH 111 will focus on chemical concepts important for students in health care, biological applications, and environmental studies to understand. Topics include atomic structure, solutions, acids, bases, stoichiometry, equilibrium, organic functional groups, and biomolecules. It would be a mistake to assume that all the material covered in the lectures appears in the text, or vice versa. Therefore, it is important to study both the text and the lecture notes.

Course Learning Goals:
Participating in this course will allow students to begin developing their ability to:
• Describe matter and transformations of matter at an atomic level
• Use representations and models to describe, understand, and predict chemical phenomena applicable to everyday situations
• Think critically and solve quantitative word problems
• Prepare and evaluate claims based on scientific evidence
Course Expectations:
Students enrolled in CH 111 are required to learn a new language - the language of chemistry. This class will focus on understanding and describing chemical phenomena in terms of macroscopic, sub-microscopic, and symbolic representations. Success in this course requires not only an understanding of the basic vocabulary, facts and concepts, but also the ability to critically analyze relationships between phenomena and to apply knowledge to novel situations. In addition to presentations by the instructor, students will participate in classroom activities and assignments designed to promote development of problem-solving and thinking skills required for success in University level science classes. Students are expected to prepare for and attend class, arrive on time and not leave early, read and study the assigned sections in the textbook, complete assignments, participate in classroom activities and discussions by working in groups, and ask questions when you need help. Please respect fellow students and refrain from chatting, cell phone use, outside reading, web surfing, gaming and text messaging during class. In return, your instructor will guide your learning, help you develop a framework for your new knowledge, and facilitate how to apply what you have learned to new situations.

Grading:
Final grades will be assigned based on the following:

<table>
<thead>
<tr>
<th>Component</th>
<th>Percentage</th>
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<tbody>
<tr>
<td>Pre-course assessment</td>
<td>1%</td>
</tr>
<tr>
<td>Online homework</td>
<td>10%</td>
</tr>
<tr>
<td>(10 assignments of equal weight)</td>
<td></td>
</tr>
<tr>
<td>Weekly quizzes (5)</td>
<td>10%</td>
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<tr>
<td>(6 quizzes, lowest score dropped)</td>
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</tr>
<tr>
<td>Midterm 1 (Wed. Oct. 19th, Week 4)</td>
<td>25%</td>
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<tr>
<td>Midterm 2 (Wed. Nov. 9th, Week 7)</td>
<td>25%</td>
</tr>
<tr>
<td>Final Exam (10:15am Fri. Dec. 9th)</td>
<td>29%</td>
</tr>
<tr>
<td>Total</td>
<td>100%</td>
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The following percentages ensure the corresponding grades: 90% = A, 78% = B, 65% = C, 50% = D, <50% = F. These cut-offs may change at the end of the term, depending on the course average. Students who choose the P/NP option must earn the equivalent of a C- to receive a P (pass) in this course. Students who receive a C- or higher grade (or P) in this course are not eligible to re-take the class at a later date. A course grade of incomplete (I) will be considered only for individual cases with cause. An incomplete grade is not mean to be a substitute for an undesirable regular letter grade. An incomplete may be issued when the quality of work is satisfactory, but some minor yet essential requirement has not been completed, for reasons acceptable to the instructor (http://registrar.uoregon.edu/incomplete_policy).

Pre-course assessment: Students must complete and submit the online ALEKS assessment by 11:50pm on Wednesday, September 28th. This assessment will take approximately 2 hours to complete. Students who complete this assessment will earn 1% towards their final grade.

Homework Assignments: Students must complete and submit online homework assignments available using Pearson’s Mastering Chemistry homework system. All assignments are due at 11:59 pm on the date indicated on this syllabus (usually Tuesdays). On-line homework assignments are worth 10% of the course grade. Study groups are encouraged. Each student is responsible for completing and submitting their own work.

Deadlines for assignments are not negotiable and technology failure is not an excuse for late work.

Flipped Fridays: Students are responsible for viewing an online presentation and/or answering questions online BEFORE attending class. During class, students will work in groups to solve problems based on the online material.
Assessment Wednesdays:

**Weekly Quizzes:** Every week, except for during Weeks 1 & 5, there will be an in-class quiz or exam. Most weeks, these assessments will take place on Wednesdays (see class schedule). During Week 9 (Thanksgiving), the quiz will be on Monday. In addition, there will be a comprehensive final exam at the end of the term. Quiz scores will be based on the highest five scores out of six quizzes. Make-up or early exams/quizzes will not be given. Students who have an authorized and unavoidable UO class or club event that conflicts with an exam or quiz should notify Professor Baldock as soon as possible.

**Clicker Registration:** Students are responsible for bringing an i>clicker2 to each class session. Students must register their clickers via the i>Clicker link on their CH 111 Canvas page by Friday, October 7th, 2016. Registrations made through the non-UO i>clicker website will not link to the University of Oregon system. Any points earned before successfully completing clicker registration can be awarded retroactively. Failure to meet the registration deadline will result in points not being awarded until the end of the term.

**Extra Credit:**

*Non-flipped classes:* Each week, students will have the opportunity to answer questions based on the reading and the material presented during class. Up to 2.0 percentage points of extra credit can be earned by submitting answers to clicker questions using an i>clicker2 (2 participation points per question).

*Friday flipped classes:* Up to 1.0 percentage points of extra credit can be earned by participating in activities related to flipped Friday classes. Extra credit will be based on responses to in-class clicker questions (2 points for incorrect answers, 5 points for correct answers). Each student may drop their lowest clicker question score.

**Calculator policy:** A *simple, inexpensive* scientific calculator is required for use during quizzes/exams. The calculator should be capable of square roots, logarithms, scientific notation operations, and have a $y^x$ key. Calculators that can be programmed, communicate with other devices, store text, product graphs, or that make noise are NOT ALLOWED. Violation of the calculator policy will result in academic sanctions.

**Exam/Quiz policies and procedures:** An exam/quiz booklet will be provided, with a seat number provided on the front. Students are required to sit in their assigned seat. Test booklets assigned to left-handed desks will be available at the front of the classroom.

1. Exams and quizzes will be administered on the days and times indicated on this syllabus. Written requests for alternate testing arrangements due to conflicts with authorized and unavoidable university events or due to a valid academic accommodation must be made during the first two weeks of the term. Documentation must be included with all requests. No make-up exams or quizzes will be offered.

2. Students must bring #2 pencils, an approved non-programmable calculator, and student identification card to quizzes and exams.

3. Only approved calculators (see Calculator policy, above) may be used during quizzes and exams. Sharing of calculators is not allowed.

4. All calculators and associated covers are subject to search and inspection before the start of and during exams and quizzes; improper calculators may be temporarily confiscated or seized. A limited number of calculators will be available to check out during the exam on a first-come, first-served basis. Use of these calculators will result in a 5% penalty on the exam.

5. The use of electronic dictionaries during exams is not allowed. Paper dictionaries must not contain ANY extra writing and must be presented to the instructor or head proctor before the beginning of the exam for inspection. The instructor and head proctor have the right to refuse a student the use of a dictionary for good cause.
6. UO Student identification cards must be brought to each lecture quiz and exam. Students are required to show their UO Student ID when requested to do so by the instructor, assisting proctor, or UO staff.

7. The "hat rule" will be enforced during all quizzes and exams. (All baseball style caps or brimmed hats must be removed. Upon approval of the instructor, for a valid medical reason, baseball style caps may be turned backwards.)

8. All cell phones (i-phone) and other wireless communication devices (i-pads) must be turned OFF and stowed out of sight before beginning a quiz or exam. Use of a cell phone during an exam for ANY reason will be regarded as a violation of academic dishonesty guidelines. Students experiencing an emergency situation should speak to the course instructor.

9. Headphones and unauthorized earpieces must not be used during quizzes and exams.

10. Students must keep all exam or quiz material at their seat – not on adjacent seats. ALL other personal materials (bags, packs, phones, etc.) must be stored under the appropriate seat during the quiz/exam.

11. A Periodic Table, appropriate equations, and physical constants will be provided on each quiz/exam.

**Academic Dishonesty:** Academic dishonesty in any guise, including plagiarism, fabrication, and cheating, will not be tolerated. All work submitted by students must be their own work and produced exclusively for this course. The use of sources (ideas, quotations, paraphrases) must be acknowledged and cited using a consistent format. Because one of the best ways to fully learn a subject is to try and teach it to another person, collaborative and group work are very much encouraged. However, students must submit their own answers homework, quiz and exam questions.

**Electronic devices:** The use of electronic devices during lecture is distracting to other students. Students who plan to use a laptop to take notes during class must sit in the upper balcony of the classroom. All uses of cell phones and other electronic devices during exam times are prohibited.

**E-mail Policy:** The University of Oregon has adopted university email addresses as an official means of communication. It is the responsibility of each individual student to regularly check their UO email account to stay current with course communications.

Email is a very handy method for communicating with people, whether around the world or across campus. In some situations email protocol is not as formal as paper communications. In others, (academics, jobs, etc.) there are some expectations of protocol. When communicating with us (or any faculty member or GTF) by email, please adhere to the following guidelines:

- The subject line should indicate the course number and the nature and topic of the email.
- Spell correctly and use appropriate punctuation.
- Use appropriate forms of salutation.
- Always sign messages with first and last name.Unsigned messages will not receive a response.

Please refer to the syllabus and/or course materials posted on Canvas for answers to questions before sending the course instructor/TA an email. If an answer to your question is readily available from one of those sources, do not expect to receive a response in a timely manner.

**Inclement Weather Policy:** Check the UO home page to determine if the university has suspended classes. In that case, no further action is required. In the unlikely event that we cancel a class due to inclement weather even though the university remains open, an announcement will be made on the CH 111 Canvas web site and an email will be sent to all students.

**Tutoring:** The Department of Chemistry and Biochemistry provides contact information for private tutors. For more information, visit the office in room 91 Klamath Hall.

**Accessible Education:** The University of Oregon is working to create inclusive learning environments. Any student with a documented disability, who may anticipate needing accommodations in this course, should
request the appropriate documentation for Professor Baldock from the Office of Accessible Education (164 Oregon Hall). The testing center fills up quickly, so requests need to be submitted early in the term.